



**Radiological data on  
building stones from  
a Spanish region**

A. Pereira

**Radiological data on building stones from  
a Spanish region: Castilla y León**

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Received: 17 July 2013 – Accepted: 26 July 2013 – Published: 14 August 2013

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Published by Copernicus Publications on behalf of the European Geosciences Union.

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## Abstract

As construction and building material, natural stone has a great potential to promote the commercial activities of certain European regions. Such is the case of Castilla y León in Spain, where many different rocks, ranging from sedimentary to metamorphic and igneous, are commercialized for building purposes. However, to be able to compete in a market subject to an economic crisis, highly exacerbated in the construction sector, and to compete with the lower prices offered by emerging countries, the issue would be to make the Spanish offer more attractive. Here we propose a complete characterization of rocks regarding their radiological properties, which are related to their mineralogy and geochemistry. Rocks emit natural radioactivity and the presence of Rn and its decay products in dwellings has become an important issue in North America and Europe owing to its relationship with the carcinogenic effects of this gas. Although most of the rocks studied are within the European Norm 112 for Radiology Protection / values accepted in the Euratom Basic Safety Standards Directive (accepted value  $I \leq 1$ ; average for sedimentary rock:  $I = 0.22$ , SD 0.14; average for metamorphic rocks:  $I = 0.70$ , SD 0.48; average for igneous rock:  $I = 0.86$ , SD 0.22), the inclusion of a proper radiological characterization in the list of characteristics would guarantee quality and safety in their use in comparison with products lacking this information. Some natural stones have been demonized for the potential exhalation of natural radioactivity and the different parameters used should be addressed in a more systematic way.

## 1 Introduction

The carcinogenic effects of radon gas on the human population have been clearly established (Lubin et al., 1994; Darby et al., 1995; Krewski et al., 2006; Chen et al., 2011) and this has prompted many countries to address the question of the presence of radon inside dwellings in their territories (Gillmore and Jabarivasal, 2010). It is also well known that radon is the largest single contributor to exposure of the public to nat-

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5 ural radiation (UNSCEAR, 1993). However, despite the large dataset already available for radon in soil, water and air, only limited work has been published in the international scientific literature on radon concentrations in natural building stone in general (Pavlidou et al., 2006) and in Spanish ornamental stone in particular (Pereira et al., 2011).

10 In this latter case the exceptions are certain widely commercialized stones (i.e. Rosa Porriño granite, from Galicia) and even in this case some data and locations have been found to be incorrect (Pereira et al., 2011).

15 Natural stone is the most important natural resource in Castilla y León, Spain (SIEM-CALSA, 2008). The sector employs more than seven thousand people directly or indirectly and a large amount of the product (up to 20 %) is exported, mainly within Europe (<http://www.pinacal.es/ingles/index.asp>, last seen on 6 August 2013). Different lithologies are found in this area, ranging from sedimentary to igneous and metamorphic rocks (Fig. 1). For many years the building sector has been an important driver of the economy, some of the quarries being used both for new construction and for restoration, since many cities in the area have been acknowledged as belonging to the UNESCO World Heritage (Pereira and Cooper, 2013) and restorations involving the replacement of slabs and ashlar must necessarily use the original local natural stone. The buildings are constructed using the local stone, mainly sandstone and limestone, but several granites are used as the bottom ashlar because of the problems arising from the capillary absorption of damp by those sedimentary stones (Schillaci et al., 2008). Some polished decorative stone slices are also present on some of the buildings. The exterior use of building stone is not a problem regarding radioactivity since radon escapes into the atmosphere. Radon exhalation is a normal occurrence, and the terrestrial atmosphere contains a low but fairly constant level of the radioactive gas. After several days of residence time in the atmosphere,  $^{222}\text{Rn}$  naturally decays to  $^{218}\text{Po}$ , a metallic radionuclide that falls back to the ground with dust and rain over a period of hours/days (Liang et al., 2004). However, the increasing practice of using natural stone as polished slabs for pavement and worktops in both kitchens and bathrooms instead of other synthetic materials, which were very popular years ago, has raised

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Isotopic activities were estimated using an Ortec NaI(Tl) detector (7.6 cm) protected from background radiation with a lead shield at the Laboratory of Natural Radioactivity of the University of Coimbra. The samples (0.5 kg) were prepared at a grain size of less than 0.5 mm, placed in 0.2 L Marinelli beakers (internal surface area of 0.0219 m<sup>2</sup>) and left for one month for equilibrium in the <sup>226</sup>Ra chain to be reached before measurements were taken. Counting was performed over 10 h and the spectra were processed with Scintivision-32 software. The background was evaluated using a Marinelli beaker filled with a radionuclide-free material (pure quartz) under the same analytical conditions, and subsequently subtracted from all samples. Calibration was achieved using standards from the International Atomic Energy Agency: RGK-1 (potassium sulphate), RGU-1 (diluted uranium ore) and RGTH-1 (diluted thorium ore) of known composition and activities. K was evaluated from the <sup>40</sup>K peak (1460.8 keV), U from the <sup>214</sup>Bi peak (1764.5 keV), and Th from the <sup>208</sup>Tl peak (2614.5 keV), assuming secular equilibrium; <sup>226</sup>Ra activity can be estimated directly from <sup>214</sup>Bi on the same assumption. Uncertainties are strongly dependent on the radionuclide concentration. The average uncertainties observed in this study were 3.6 % for <sup>40</sup>K, 11.5 % for <sup>214</sup>Bi and 36 % for <sup>208</sup>Tl.

Radon exhalation was determined from the same powder samples (previously dried in an oven for 24 h at 40 °C). Samples (0.5 kg in weight) were placed inside a 12 L capacity radon-proof container made of stainless steel for 4 weeks. The concentration of radon was then measured with an AlphaGuard Pro monitor and the exhalation rate was estimated according to the following equation:

$$Er = ((Cr \cdot V) / W) \lambda, \quad (1)$$

where  $Er$  is the exhalation rate (Bq kg<sup>-1</sup> h<sup>-1</sup>);  $Cr$  is the equilibrium radon concentration in the container (Bq m<sup>-3</sup>);  $V$  is the free volume of the container (L);  $W$  is the weight of the sample (kg), and  $\lambda$  is the radon decay constant (h<sup>-1</sup>). Uncertainties varied between 10 and 20 % for the typical range of values measured in this work.

A detailed petrographic study using a Leica microscope with a coupled digital camera was used to reproduce the most interesting features of our natural stones.

### 3 The rocks

The rocks studied are described in detail in Garcia de los Rios and Baez (2001) and here we summarize the mineralogy since this is assumed to be the main factor controlling Rn exhalation in most rocks.

#### 3.1 Sedimentary rocks

Sedimentary rocks are very important in this region and they have been used to sculpt the façade work on most historical buildings. The radiological measures in sedimentary rocks were performed on the principal rock types used in Castilla y León and neighbouring regions as building stones: siliciclastic sandstones, a conglomerate, and limestones.

##### 3.1.1 Sandstones

The sandstones studied correspond to feldspathic litharenites (Folk, 1965). In decreasing order of abundance the grains consist of quartz, rock fragments (schists, polycrystalline quartz), feldspars (orthose and plagioclase), and micas (muscovite and biotite) as the major components, with accessory minerals such as iron oxides, tourmaline, zircon and/or rutile. The matrix content commonly represents more than 30 % and is formed by small crystals of mica and clays. Some of these rocks are patchily cemented by iron oxides and barite.

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### 3.1.2 Conglomerate

We included a conglomerate that has been quarried since Roman times and that has been used for important historical buildings in Spain. The peculiarity of this conglomerate is that it is fully cemented by a mosaic of calcite and is known as Mármol de Espejón (Espejón Red Marble).

### 3.1.3 Limestones

Limestones are quarried both for new constructions and for the restoration of historical buildings. The main types are dolomicrites (i.e. Silos stone) and biocalcarenites with abundant fossils (i.e. Hontoria limestone, Campaspero limestone) (Folk, 1959).

## 3.2 Igneous and metamorphic rocks

The igneous (granites and granitic rocks) and metamorphic (phyllite and gneiss) rocks analysed in this work cover the whole variety present in the region.

Within the granitic rocks, the most common types are monzogranites and several leucogranites and granodiorites, made up of quartz, potassium feldspar, plagioclase, different micas, possible amphibole and different accessory phases. All of them have biotite, and muscovite is present in more evolved and aluminium-rich rocks, where cordierite can also be found. Hornblende appears in minor amounts in several granodiorites. These granitic rocks show diverse microstructures, from fine-grained equigranular types to porphyritic facies, where large crystals of potassium feldspar are found.

From a petrographic and petrogenetic point of view, the granitic rocks studied belong to different families. One of them was mainly formed by the melting of metasediments and the other is a hybrid type, formed by a mixture of sources. The former contains accessory minerals such as zircon, xenotime and monazite. U and Th are present in their mineral lattice. In addition to the above accessories, the latter contains allanite (Fig. 2), epidote, and sometimes thorite. The formula of allanite reveals the presence

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of a significant amount of rare-earth elements, some of them highly radioactive, plus some U and Th.

One particular case is Rojo Sayago, a porphyritic granite transformed into epysienite (Le Maitre et al., 1989), where calcic plagioclase has been converted into albite, biotite into chlorite and the iron contained within the potassium feldspar has been oxidized. All these changes are due to local low-temperature hydrothermal metamorphism. Zircon appears mostly within other mineral phases, although its accumulation has also been observed in deformation structures (Fig. 3a and b).

The gneiss from the province of Zamora belongs to the so-called “Ollo de Sapo” formation, a previous igneous rock that has been deformed and recrystallized (Castro et al., 2000). It shows a gneissic structure, is medium- to coarse-grained and it is characterized by potassium feldspar crystals as large as 3–4 cm, surrounded by a foliation formed of muscovite and biotite. Other porphyroclasts are albite-oligoclase and bluish quartz; accessory minerals such as zircon, titanite, ilmenite and monazite are common. The accessory minerals are usually related to biotite, forming micronodules with it.

The phyllite is a metasedimentary rock of Neoproterozoic age. It is fine- to very fine-grained, showing a well-developed schistosity. It is made up of phyllosilicates (e.g. phengite, chlorite and minor biotite), and contains apatite, tiny zircon crystals and ilmenite as accessory minerals.

## 4 Results

In a previous study involving 20 samples, each of them related to a different ornamental stone, Manteca (2011) established the possible use of the rocks on the basis of their radiological risk by determining the  $I$  index proposed in Norm 112. As expected, igneous and metamorphic rocks showed similar averages, being the highest of the global dataset, but they also showed differences as regards to variance, which was much higher in the metamorphic group than in the igneous type (Table 1). Lithology should be the main factor controlling variability since in the latter case granites were

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the only rock type present (coefficient of variation of 0.26) while the metamorphic group comprised granite-related rocks, such as gneisses, but also a phyllite and a quartzite; this last rock type had the minimum value ( $I = 0.16$ ) while the highest ( $I = 1.08$ ) was measured in the gneissic rock. These values are slightly higher than the level demanding action in Norm 112, and the same conclusion applies for two samples of the igneous group (the highest value of  $I = 1.24$ ; Manteca, 2011).

The global average for the sedimentary rocks ( $I = 0.22$ ) was lower than for those of the igneous and the metamorphic group, but showed the highest variance in the dataset (coefficient of variation of 0.65). This variance was also related to the lithology, composed of sandstones, limestones and a conglomerate. Accordingly, the rock composition, and the U, Th and K content are the main factors that control the  $I$  parameter.

The results of the  $^{226}\text{Ra}$ ,  $^{232}\text{Th}$  and  $^{40}\text{K}$  activities in the selected set of 17 samples are shown in Table 2. The  $^{222}\text{Rn}$  exhalation rate obtained for each sample is also included in the same table. Owing to the low number of samples in the metamorphic group ( $n = 2$ ) and because the results were similar to those observed in the igneous rocks, both rock types are discussed together.

The isotope activities and radon exhalation rate displayed the same pattern, characterised by lower values for the sedimentary samples and higher ones for the igneous and metamorphic group. Sandstones, limestones and the conglomerate show varying  $^{226}\text{Ra}$ , between less than the detection limit for the technique used up to  $27.1 \text{ Bq kg}^{-1}$ , with an average of  $15.1 \text{ Bq kg}^{-1}$ . In this group the lowest values were found in the carbonate rocks and in the conglomerate, as expected; the sandstones showed the highest activities. For the same rock types,  $^{232}\text{Th}$  and  $^{40}\text{K}$  showed a similar variability, with a maximum of  $25.2 \text{ Bq kg}^{-1}$  and  $749.5 \text{ Bq kg}^{-1}$ , respectively. The estimated average for  $^{232}\text{Th}$  was  $9.5 \text{ Bq kg}^{-1}$  and  $301.9 \text{ Bq kg}^{-1}$  for  $^{40}\text{K}$ . The radon exhalation rate was also variable, between less than the detection limit up to  $0.017 \text{ Bq}\cdot\text{m}^{-2}\text{h}^{-1}$ , with an average of  $0.005 \text{ Bq}\cdot\text{m}^{-2}\text{h}^{-1}$ . As before, the lowest values were measured in the carbonates.

The igneous-metamorphic group showed higher values for all the parameters analysed. The  $^{226}\text{Ra}$  activity ranged from  $39.0 \text{ Bq kg}^{-1}$  to  $118.4 \text{ Bq kg}^{-1}$ , with an average

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is of 3.3 and 2.2, respectively), while Rojo Sayago is depleted in the same radioisotope ( $^{226}\text{Ra}/^{232}\text{Th}$  is of 0.6). This latter rock, reddish in colour, shows evidence of the hydrothermal alteration that transformed biotite into chlorite and plagioclase into carbonates and epidote, dissolving the quartz and leaving a high percentage of porosity. In fact, the proper terminology for this rock should be “epysienite” instead of granite, and the hydrothermal alteration can explain the removal of the uranium, probably carried out by hot fluids. For the other granites, the explanation for the observed geochemical behaviour could be related to the presence of uranium-bearing minerals.

The correlation of  $^{226}\text{Ra}$  activity with radon exhalation also shows a linear trend, but with some samples deviating from such linearity (Fig. 7). Three of them, all included in the igneous and metamorphic group, differ significantly due to an increased radon exhalation rate for equivalent radioisotope activity. This may be related to the mineralogical distribution of  $^{226}\text{Ra}$  (and uranium) in the rocks. The highest radon exhalation rate was measured in the Rojo Sayago reddish granite, which allowed us to conclude that despite the lower  $^{226}\text{Ra}$  activity (and uranium content, Manteca, 2011) the changes in rock mineralogy (imposed by hydrothermal fluids and the tectonics, increasing the porosity and leading to a probable redistribution of the uranium-chain isotopes in the rock) induced a strong increase in the ability of the rock to emanate radon gas. It is highly likely that a redistribution of the long-lived radiosotopes of the uranium decay chain also occurred in the mineralogy of Rubio Cardeñosa sample since this also shows a greater increase in the  $^{222}\text{Rn}/^{226}\text{Ra}$  ratio than expected. Tectonic deformation may be another key factor influencing the increase in the  $^{222}\text{Rn}/^{226}\text{Ra}$  ratio, as observed in the case of the Zamora Gneiss sample.

By contrast, the Silvestre Dorado sample revealed the opposite trend, showing an unexpected decrease in the above ratio, with a radon exhalation rate lower than that measured, for instance, in the Rubio Cardeñosa ( $0.03\text{ Bq kg}^{-1}\text{ h}^{-1}$  vs.  $0.07\text{ Bq kg}^{-1}\text{ h}^{-1}$ , respectively). While the latter showed the lowest  $^{226}\text{Rn}$  activity in the igneous and metamorphic group ( $39\text{ Bq kg}^{-1}$ ), the Silvestre Dorado, in the same group, had the highest activity ( $118.4\text{ Bq kg}^{-1}$ ). Rocks that have a high radon exhalation rate do not neces-

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sarily exhale dangerous amounts of the gas. (<http://www.sustainable-design.ie/arch/radon.htm>, last seen 19 November 2012). This is always been poorly understood by the uninformed public, producing fear, and should be explained in detail.

Fission-track techniques have been used to study the microscopic distribution of uranium in rocks (Kleeman and Lovering, 1967) and these revealed that the above geological processes can effectively induce a redistribution of uranium in the rocks, increasing their ability to promote radon gas exhalation exponentially. In contrast, the confinement of the same chemical element to the structural lattice of the minerals significantly reduces the escape of this gas.

The European Radiology Protection Norm 112 describes the maximum value of radiation that building materials can emit with the *I*-index, calculated from the  $^{40}\text{K}$ ,  $^{232}\text{Th}$  and  $^{226}\text{Ra}$  activities. A few European countries follow this norm as a safety factor for construction requirements, but in most of them it is not considered. It should be recalled that *I* is calculated in the scenario of the integral use of the rock (walls, floors and ceilings).

The *I*-index proposed in Norm 112 shows values  $< 1$  (Fig. 8) for most of our cases, which means that the rocks can be used for any type of building purposes (El-Hussein, 2004). The correlation of the *I*-index with the radon exhalation rate again reveals the positive linear trend between the two variables, which should indicate, as a general rule, the correct applicability of the *I*-index as an estimator of the radiological risk, including radon gas exposure, associated with the use of the rocks studied as building materials.

However, the results obtained in the present study show some deviations from the general trend that deserve more detailed analysis. Despite the recognition of Rojo Sayago and Zamora gneiss as non-recommended for indoor use, the *I*-index is not proportional to the radiological risk derived from the measured radon exhalation rate. Despite this, the parameter is higher than 1.0 in the case of the Silvestre Dorado sample but the measured radon exhalation rate does not differ from the variation interval observed in rocks from the same lithological group, with an *I*-index below the action

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limit. Moreover, the anomalous radon exhalation rate seen for the Rubio Cardeñosa sample seems to be incorrectly reflected by the  $I$ -index.

The red Rojo Sayago is used only in exterior applications because the mineralogical and textural characteristics of this rock, induced by low temperature hydrothermal alteration and deformation, do not permit it to be used as polished ornamental stone indoors; outside, the exhaled radon would disperse into the surrounding atmosphere. Unfortunately, this is not the recommendation found in commercial catalogues and manuals (García de los Ríos and Baez, 2001). Owing to its physical and petrologic characteristics, the gneiss is used as a natural ornamental stone on façades and in paving, with bush hammering and natural endings only on the outside. One particularity of this rock is that it is being exported to France, where the “gneiss de Limoges” is very popular but the quarries are almost exhausted for stone walls. In this case, because the rock is only used outside, the higher  $I$  value is harmless. However, in all cases customers should always be informed.

As expected, granites and related lithologies (gneiss) show the highest average U concentrations (Table 2), although within the regular values for such rocks (Rogers and Ragland, 1961; Pavlidou et al., 2006) and without any significant anomalous enrichment.

## 6 Conclusions

Most of the natural stones from Castilla y León studied in this work show  $I$  values of less than 1, which means that they can be used for any type of building purpose. Some of them have values close to zero, these cases being from the sedimentary lithology group. Two samples have higher  $I$  values: one is a hydrothermally transformed granite ( $I = 1.24$ ) and the other is a gneiss ( $I = 1.08$ ). However, it is currently uncommon to find entire rooms (ceiling, walls and flooring) made of the same material (mainly because none of these natural stones admits good polishing) and in any case we have observed that these rocks are only used for exterior purposes, not because of their radioactivity

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results with us to complete the data set of the selection. A. Ferrero carried out her final project under the guidance of A. Pereira and D. Pereira, sharing her results with us as well. Nic Skinner is appreciated for his help in polishing English grammar. Bruce D. Malamud is acknowledged for his suggestions to improve the manuscript.

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**Table 1.** / values for natural stones from Castilla y León (Manteca, 2011).

	Sedimentary	Metamorphic	Igneous
Average	0.22	0.70	0.86
Median	0.22	0.86	0.87
Standard deviation	0.14	0.48	0.22
Coefficient of variation	0.65	0.69	0.26
Minimum	0.01	0.16	0.59
Maximum	0.47	1.08	1.24
<i>N</i>	14	3	9

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**Table 2.** U,  $^{226}\text{Ra}$ ,  $^{232}\text{Th}$ ,  $^{40}\text{K}$  and Rn exhalation in the powdered samples of natural stone studied (only for the 17 selected samples). Commercial names have been kept in Spanish for easier referencing in the original literature. # is location in Fig. 1.

#	Commercial name	Lithology	U (ppm)	$^{226}\text{Ra}$ ( $\text{Bq kg}^{-1}$ )	$^{232}\text{Th}$ ( $\text{Bq kg}^{-1}$ )	$^{40}\text{K}$ ( $\text{Bq kg}^{-1}$ )	Exhalation rate ( $\text{Bq kg}^{-1} \text{h}^{-1}$ )
1	Arenisca Quintanar	Sandstone	1.7	21.3	7.8	485.2	0.002
2	Quintanar beige	Sandstone	0.7	9	9.1	332	0.004
3	Rojo San Adrián	Sandstone	2.2	27.1	25.2	749.5	0.009
4	Sierra Demanda	Sandstone	1.8	22.6	6	410.1	0.005
5	Zamora	Sandstone	0.8	23.4	23.7	438.7	0.017
6	Espejón	Conglomerate	0.5	6.2	2.9	0	0.003
7	Silos	Limestone	0.9	11.5	0	0	0.000
8	Hontoria	Limestone	0	0	1.3	0	0.000
9	Filita Bernardos	Phyllite	4.5	55.7	65.3	1016.3	0.017
10	Gneis Zamora	Gneis	6.3	77.6	68.8	1421.2	0.122
11	Rojo Sayago	Epysienite	4.3	53.6	86.2	1861.3	0.395
12	Azul noche	Granite	4.8	59.7	50.8	954.1	0.002
13	Rubio Cardeñosa	Granite	3.2	39	11.7	1201.1	0.066
14	Silvestre Dorado	Granite	9.6	118.4	53.3	1335	0.030
15	Los Santos	Granite	6.8	75.5	71.9	1123.4	0.044
16	Gris Villa	Granite	5.2	64.8	68.8	919.6	0.024
17	Sorihuela	Granite	6.1	75.3	63.6	994.1	0.012

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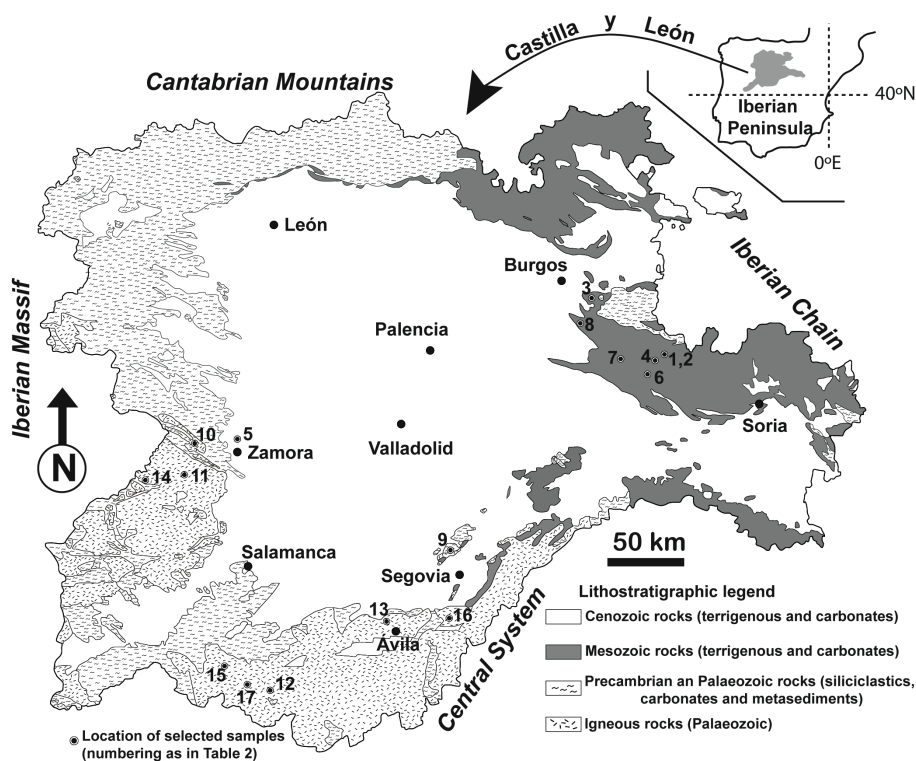
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**Fig. 1.** Simplified geological map of Castilla y León, showing the disposition of all lithologies studied in the present paper. Modified from Cabrera Ceñal et al. (1997).

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**Fig. 2.** Allanite( $\text{Ce, Ca, Y, La}$ )<sub>2</sub>( $\text{Al, Fe}^{+3}$ )<sub>3</sub>( $\text{SiO}_4$ )<sub>3</sub>( $\text{OH}$ ) in the Cardenosa granite. Objective  $\times 10$ , crossed nicols.

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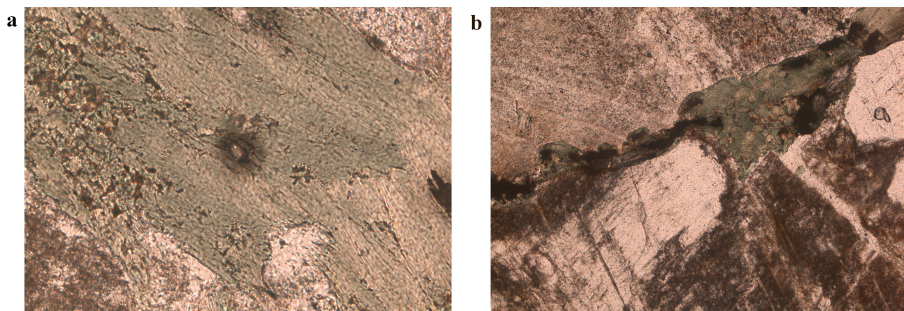
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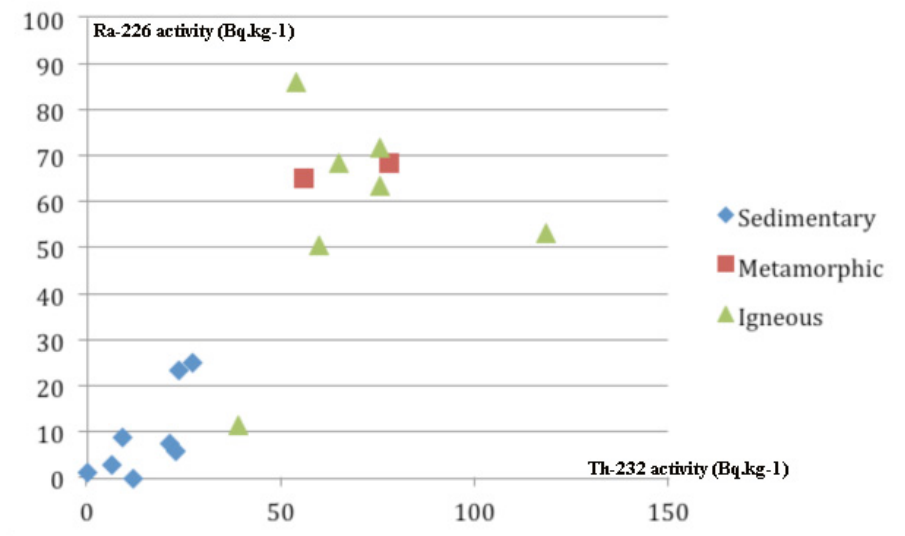


**Fig. 3.** (a) Zircon in chlorite; (b) Zircon in deformation structure. Objective  $\times 10$ , parallel nicols.

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**Fig. 4.** <sup>226</sup>Ra and <sup>232</sup>Th activity for the studied samples, in Bq kg<sup>-1</sup>.

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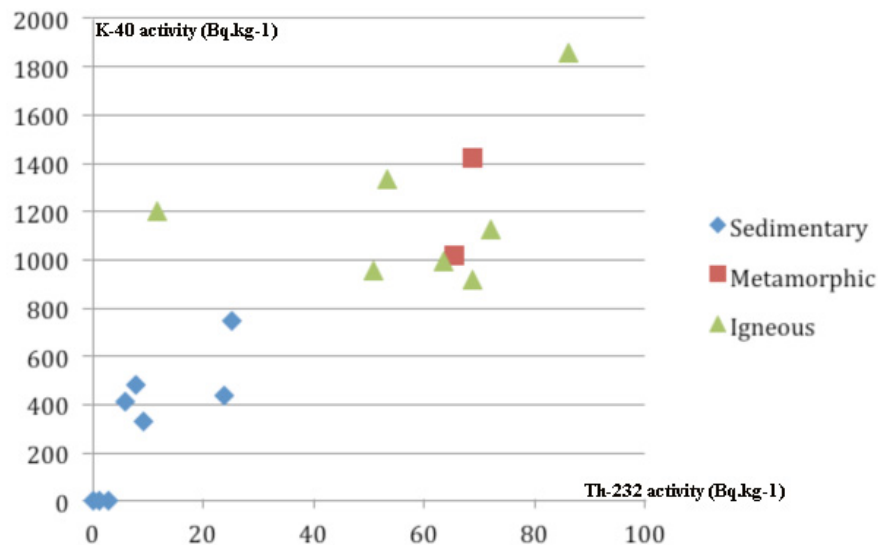
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**Fig. 5.**  $^{232}\text{Th}$  and  $^{40}\text{K}$  activity for the studied samples, in  $\text{Bq kg}^{-1}$ .

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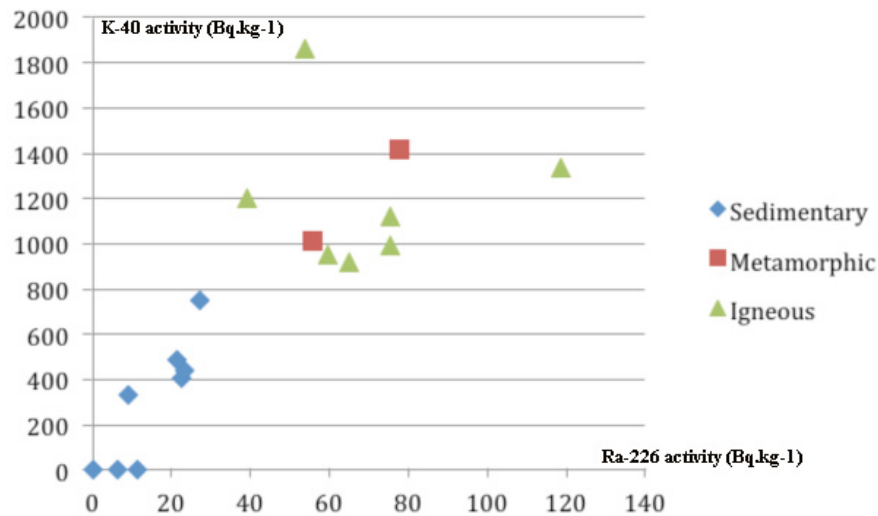
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**Fig. 6.** <sup>226</sup>Ra and <sup>40</sup>K activity in Bq kg<sup>-1</sup> for the studied samples.

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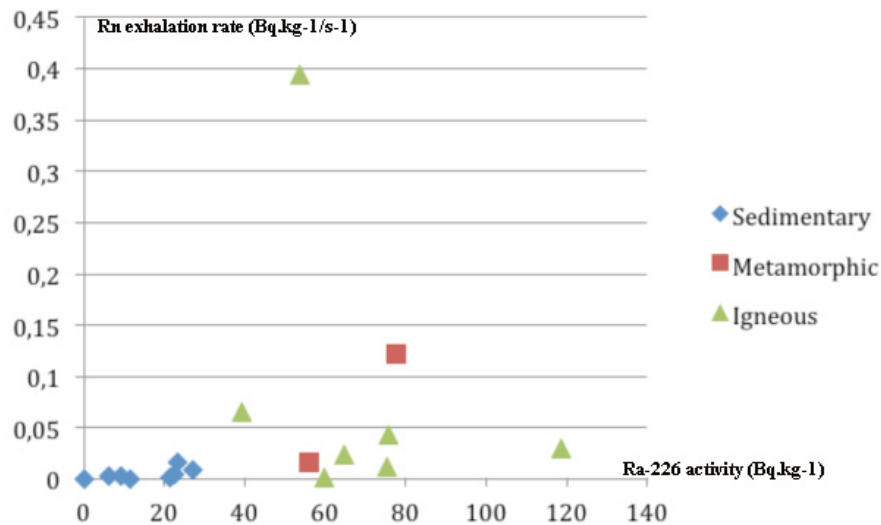
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**Fig. 7.**  $^{226}\text{Ra}$  activity in  $\text{Bq kg}^{-1}$ , and radon exhalation rate in  $\text{Bq kg}^{-1} \text{ s}^{-1}$  for the studied samples.

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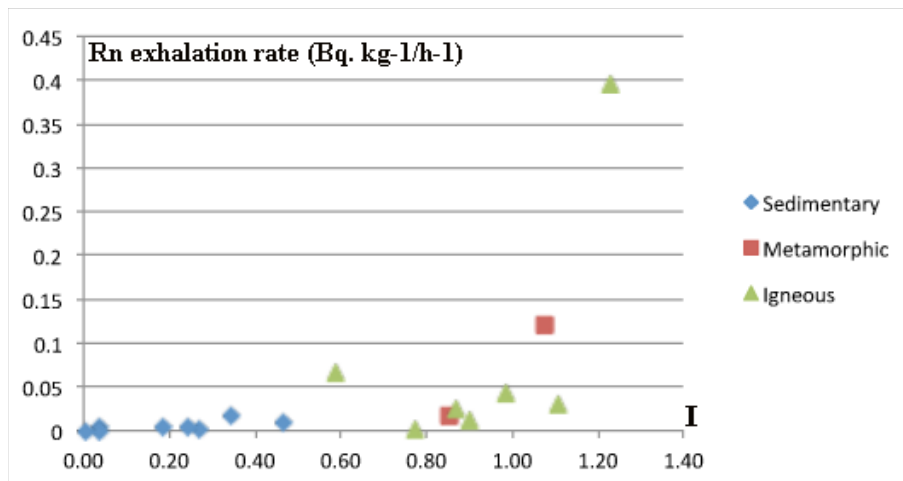
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**Fig. 8.** *I*-parameter, dimensionless, calculated from the measured isotopes and the radon exhalation rate, in  $\text{Bq kg}^{-1} \text{h}^{-1}$  for the studied samples.

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